

## THE GNCS FACTSHEETS

# Mitigating SF<sub>6</sub> Emissions

Sulfur hexafluoride (SF<sub>6</sub>) is a powerful greenhouse gas due to its highly effective absorption of infrared radiation as well as its extremely long atmospheric life span.<sup>1</sup> Therefore, despite its small atmospheric concentration relative to carbon dioxide (CO<sub>2</sub>), SF<sub>6</sub> contributes a disproportionately large share to anthropogenic radiative forcing.<sup>2</sup> Although emissions of SF<sub>6</sub> have stayed relatively constant since 1998, their long atmospheric lifetimes (between 1,000 and 5,000 years) and the resulting accumulation of SF<sub>6</sub> in the atmosphere means that the gas's contribution to radiative forcing has increased 36% over the same period.<sup>3</sup>

Emissions from the major sources of SF<sub>6</sub> production (magnesium die casting and the manufacture and use of electrical equipment) are expected to nearly double between now and 2030.<sup>4</sup> Growth in global SF<sub>6</sub> emissions will be driven primarily by growth in developing nations through 2020: SF<sub>6</sub> emissions from electrical equipment will rise with the growth of electricity infrastructure as demand for electricity rises.<sup>5</sup> In magnesium production, China represents the largest share of the international market, driven by low-cost exports and closures of plants in the US, Canada, Norway and Japan.<sup>6</sup>

## Sources of SF<sub>6</sub>

The most significant sources of SF<sub>6</sub> emissions are from the manufacture, operation and disposal of electrical equipment (27 million tons of CO<sub>2</sub> equivalent (MtCO<sub>2</sub>e) in 2000,<sup>7</sup> representing 80% of global sales of SF<sub>6</sub>)<sup>8</sup> and

magnesium die casting (9 MtCO<sub>2</sub>e in 2000).<sup>9</sup> In addition, smaller amounts of SF<sub>6</sub> are used in semiconductor production, as a tracer gas, for sound proofing in windows, in aluminum production and as an alternative to air in filling car tires.<sup>10</sup>

An inert gas, SF<sub>6</sub> is used as an insulator for electrical equipment, to protect it from overloads and short circuit currents, as well as to connect and disconnect networks.<sup>11</sup> In this context, SF<sub>6</sub> is emitted primarily through leaks over the lifetime of the equipment (approximately one third of emissions);<sup>12</sup> during maintenance, when the equipment must be opened for repairs and SF<sub>6</sub> escapes; or when the equipment is decommissioned (maintenance and decommissioning together represent approximately two thirds of emissions).<sup>13</sup> Although SF<sub>6</sub> emissions from electrical equipment decreased significantly between 1990 and 2000 due to the high price of SF<sub>6</sub> and technical improvements,<sup>14</sup> emissions increased by 55% between 2000 and 2003 due to increased sales and higher rates of retirement for electrical equipment.<sup>15</sup>

Magnesium production, an energy intensive process, has increased worldwide due to high demand for the lightweight metal, especially in the transportation industry.<sup>16</sup> In magnesium die casting, SF<sub>6</sub> is used as a cover gas to protect the molten metal from oxidation or ignition.<sup>17</sup> It is emitted due to its function as a cover gas: SF<sub>6</sub> emissions in the magnesium industry are therefore roughly equal to the industry's consumption of the gas.<sup>18</sup>

<sup>1</sup> EPA. (2006). *Global Mitigation of Non-CO<sub>2</sub> Greenhouse Gases*. Washington D.C.: US Environmental Protection Agency. p. 14.

<sup>2</sup> The major fluorinated greenhouse gases (HFCs, PFCs and SF<sub>6</sub>) comprised only 1% of global emissions in 2000. *Ibid.* p. 12; and Schaefer, D., Godwin, D. and Harnisch, J. (2006). "Estimating Future Emissions and Potential Reductions of HFCs, PFCs & SF<sub>6</sub>." *The Energy Journal*. p. 64.

<sup>3</sup> IPCC AR4 WG1. (2007). "Changes in Atmospheric Constituents and in Radiative Forcing." Ch.2 in *Climate Change 2007: Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*. Cambridge, UK: CUP. p. 141 & 145.

<sup>4</sup> IPCC AR4 WG3. (2007). "Industry." Ch. 7 in *Climate Change 2007: Contribution of Working Group III to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*. CUP. p. 454, 471.

<sup>5</sup> EPA (2006), p. IV 188.

<sup>6</sup> *Ibid.* p. IV 205.

<sup>7</sup> IPCC AR4 WG3 (2007), p. 471.

<sup>8</sup> EPA (2006), p. IV 187.

<sup>9</sup> IPCC AR4 WG3 (2007), p. 464.

<sup>10</sup> EPA (2006), p. I 5; Schaefer, Godwin and Harnisch (2006), p. 64; and Harnisch, J. and Schwarz, W. (2003). *Cost and the impact on emissions of potential regulatory framework for reducing emissions of hydrofluorocarbons, perfluorocarbons, and sulphur hexafluoride*. European Commission, Directorate of the Environment. p. 12.

<sup>11</sup> Wartmann, S. and Harnisch, J. (2005). *Reductions of SF<sub>6</sub> Emissions from High and Medium Voltage Electrical Equipment in Europe*. Nuerenberg, Germany: Ecofys. p. 5.

<sup>12</sup> EPA (2006), p. IV 190.

<sup>13</sup> *Ibid.*

<sup>14</sup> Wartmann and Harnisch (2005), p. 25.

<sup>15</sup> EPA (2006), p. IV 188.

<sup>16</sup> IPCC AR4 WG3 (2007), p. 464.

<sup>17</sup> Harnisch and Schwarz (2003), p. 24.

<sup>18</sup> Harnisch and Schwarz (2003), p. 24; and EPA (2006), p. IV 205.

## Abatement Options

*Electrical Equipment:* SF<sub>6</sub> emissions resulting from leaks in electrical equipment can be addressed through leak detection and repair (LDAR) and, for larger leaks, refurbishment. LDAR measures have the potential to reduce approximately 30% of emissions from equipment with uncontrolled emissions.<sup>19</sup> Emissions during maintenance can be addressed by switching to equipment that requires less frequent maintenance (e.g. state-of-the-art sealed pressure systems, which generally do not require maintenance during their 40 year lifetimes)<sup>20</sup> and through training of staff, best practices for gas handling and awareness-raising (i.e. staff education and labeling programs). These contributed to a decline in SF<sub>6</sub> emissions in Europe and Japan from 1995 to 2003, despite a 60% increase in production in that period.<sup>21</sup>

Theoretically, 100% of the SF<sub>6</sub> emitted during the decommissioning of electrical equipment could be recovered, purified and reused, presenting a significant mitigation opportunity.<sup>22</sup> Practical recovery rates are closer to 80% due to increasing marginal recovery costs.<sup>23</sup> In addition, a comprehensive decommissioning infrastructure is required for large-scale gas recovery, including highly-trained staff and the legal considerations around international waste transportation.<sup>24</sup>

*Magnesium production:* Technical improvements and more efficient operational practices, which resulted from rising SF<sub>6</sub> prices between 1995 and 2000, mean the amount of SF<sub>6</sub> used per ton of magnesium produced has already reached an effective limit.<sup>25</sup> Therefore, further emission reductions must be achieved through the use of an alternative cover gas.<sup>26</sup> SF<sub>6</sub> replaced sulfur dioxide (SO<sub>2</sub>) as a cover gas in magnesium production in the 1970s.<sup>27</sup> Although the toxicity of SO<sub>2</sub> increases handling difficulties, it is commercially available, can be used in a wide range of magnesium casting processes, and poses few additional problems or costs compared to SF<sub>6</sub>.<sup>28</sup> HFC-134a is a problematic alternative, since it is also a

powerful greenhouse gas. It may, however, be preferable to SF<sub>6</sub> due to its smaller atmospheric lifetime and potency, as well as its more thorough decomposition.<sup>29</sup>

## Mitigation Costs

Multiple analyses find SF<sub>6</sub> recovery or replacement measures to be low cost and even cost saving. The US Environmental Protection Agency (EPA), for example, assesses the recovery and recycling of SF<sub>6</sub> as a cost saving measure, since recycling of the gas reduces the facility's consumption and therefore its operating costs.<sup>30</sup> The EU Environment Directorate finds that SF<sub>6</sub> replacement with SO<sub>2</sub> in magnesium die casting is "economic" for both new and retrofitted plants (retrofitting is more expensive than installing a new SO<sub>2</sub> plant).<sup>31</sup>

## Current Policies and Standards

The EPA has launched a number of voluntary partnerships with industries that emit SF<sub>6</sub>, which has resulted in approximately 11.5 MtCO<sub>2</sub>e of emissions avoided between 1990 and 2005. The EPA Partnership for Electric Power Systems represents close to half the US industry, and recorded a 45.6% decrease in SF<sub>6</sub> emissions between 1999 and 2006, primarily through LDAR, equipment upgrades and refurbishment, gas recovery and recycling, and employee training.<sup>32</sup> The EPA partnership with the International Magnesium Association (IMA), which represents around half of global magnesium production, was launched in 1999 and resulted in a 40% reduction in emissions intensity between 1999 and 2002, primarily through the optimization of equipment design and improved gas management.<sup>33</sup> The IMA has a stated goal of eliminating SF<sub>6</sub> emissions, which will necessitate the use of alternative cover gases.<sup>34</sup> World Semiconductor Council (WSC) members, which represent 90% of production, achieved a 10% greenhouse gas reduction on 1995 levels by sharing information and best practices.<sup>35</sup>

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Further resources are available at [www.theGNCS.org](http://www.theGNCS.org)

<sup>19</sup> EPA (2006), p. IV 193.

<sup>20</sup> Wartmann and Harnisch (2005), p. 7.

<sup>21</sup> IPCC AR4 WG3 (2007), p. 471.

<sup>22</sup> EPA (2006), p. IV 192.

<sup>23</sup> *Ibid.*

<sup>24</sup> Wartmann and Harnisch (2005), p. 26.

<sup>25</sup> EPA (2006), p. IV 205.

<sup>26</sup> Harnisch and Schwarz (2003), p. 26.

<sup>27</sup> *Ibid.* p. 24.

<sup>28</sup> *Ibid.* p. 30.

<sup>29</sup> *Ibid.* p. 75.

<sup>30</sup> EPA (2006), p. IV 192.

<sup>31</sup> Harnisch and Schwarz (2003), p. 30.

<sup>32</sup> EPA (2007), p. 2, 4. See: <http://www.epa.gov/electricpower-sf6>.

<sup>33</sup> See: <http://www.epa.gov/highgwp1/magnesium-sf6/basic.html>.

<sup>34</sup> IPCC AR4 WG3 (2007), p. 464.

<sup>35</sup> See: <http://www.semiconductorcouncil.org/activities>.